

CONCERNING THE IMPORTANCE OF DUAL  $\rho_{\pi}$  CONCEPT IN THE LSFE  
TREATMENT FOR THE SUBSTITUENT EFFECTS ON BASICITIES OF PYRIDINES  
AND PYRIDINE N-OXIDES

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The relative dissociation constants of both meta and para substituted pyridinium ions in  $H_2O$  at  $25^{\circ}C$  were collected from a number of sources. To understand the actual substituent effects with a combination of inductive(I) and pi-electronic(Pi) effect components, the LSFE Eq. 1 has been employed statistically for each of meta and para series:

$$\log k/k_O = \rho_i \sigma_i + \rho_{\pi}^{+} \sigma_{\pi}^{+} + \rho_{\pi}^{-} \sigma_{\pi}^{-}, \quad (1)$$

where  $\sigma_i = 0.74\sigma'$ ,  $\sigma_{\pi}^{+} = 0.415(\sigma^{+} - \sigma^O)$ ,  $\sigma_{\pi}^{-} = 0.73(\sigma^{-} - \sigma^O)$ ,

and then,  $\sigma_p^O = \sigma_i + \sigma_{\pi}^{+} + \sigma_{\pi}^{-}$ ,  $\sigma_p = \sigma_i + 1.65\sigma_{\pi}^{+} + \sigma_{\pi}^{-}$ ,

$\sigma_p^{+} = \sigma_i + 3.41\sigma_{\pi}^{+} + \sigma_{\pi}^{-}$ , and  $\sigma_m = 1.17\sigma_i + 0.50\sigma_{\pi}^{+} + 0.50\sigma_{\pi}^{-}$

on benzene reactivities. The statistical application of Eq. 1 to the present pyridine sets afforded the following results.

$$(\Delta pK_a)_{para} = 6.40\sigma_i + 9.20\sigma_{\pi}^{+} + 2.39\sigma_{\pi}^{-} - 0.03;$$

$R = 0.9965$ ,  $s = \pm 0.228$ ,  $n = 15$  (Me, Et, Cl, Br, MeO,  $NH_2$ ,  $NMe_2$ , Ph, H,  $CF_3$ , COMe, CN,  $NO_2$ , CPh, COOMe).

$$(\Delta pK_a)_{meta} = 7.60\sigma_i + 3.02\sigma_{\pi}^{+} + 2.25\sigma_{\pi}^{-} - 0.03;$$

$R = 0.9973$ ,  $s = \pm 0.149$ ,  $n = 14$  (Me, Et, F, Cl, Br, MeO,  $NH_2$ ,  $NMe_2$ , H, COMe, CN,  $NO_2$ , CPh, COOMe).

The important features of the results are (i) the ratio of inductive component for meta to para substituents,  $\rho_{i,meta}/\rho_{i,para}$ ,

is 1.19, which is virtually identical with 1.17 in the corresponding benzene system, and (ii) in the  $\pi$  effect, the contribution of  $\pi$ -acceptor substituents at para position ( $\rho_{\pi}^{-}$ ) is very small relative to that of either  $\rho_{\pi}^{+}$  or  $\rho_i$ , but definitely non-negligible: we note here that the apparent large  $\rho_{\pi}^{+}/\rho_{\pi}^{-}$  (ca. 4) does not mean a large resonance exaltation of  $\pi$ -donor substituents classified as  $\sigma^{+}$ -type reactivity.

From these results, it is clear that the I effect in pyridine system is position-dependent, and the I effect from meta position is 1.2 times more effective than that from para one. By elimination of the I effect using its contribution factor of 0.85 (=1/1.19), a  $\pi$ -electronic effect equation should be constructed for the combined data set of meta and para substituents, without the use of any inductive parameters:

$$(\Delta pK_a)_{\text{para}} - 0.85(\Delta pK_a)_{\text{meta}} = 6.6\sigma_{\pi}^{+} + 0.5\sigma_{\pi}^{-}.$$

This illustrates that in the  $\pi$  effect, there exists a bilinear relationship for  $\pi$ -donor and  $\pi$ -acceptor substituents bisected at the unsubstituted point, and there is not any other existing scales of single substituent constants with which this characteristically unbalanced  $\pi$  effect can be accounted for.

Further, the substituent effects on basicities of pyridine N-oxides have been well treated in the similar manner.

Combined with earlier results by Taft's DSP and other relevant treatments, the soundness of the dual  $\rho_{\pi}$  approach and the superiority of the LSFE treatment have been demonstrated.